



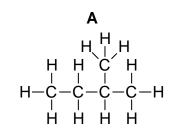
Name: ..... ( )

# Chem!stry class:

Date: ..... / ..... / .....

## **Assignment on Polymer Chemistry Multiple Choice Questions**

1. Which one of the following formulae represents a compound likely to undergo addition polymerisation?



2. Part of a polymer molecule has the following structure:

From which substance is this polymer made?

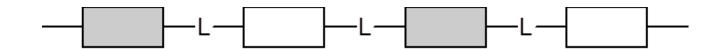
Α CH<sub>4</sub> В C<sub>2</sub>H<sub>4</sub>

C  $C_2H_6$ 

- D C<sub>3</sub>H<sub>6</sub>
- 3. The macromolecules of proteins, fats and carbohydrates can all be broken down into their simple units by a similar process. What is the process called?
  - Α Dehydration.

В Hydrolysis.

C Esterification. D Oxidation. 4. The diagram below represents the structure of a polyamide (such as nylon):



Which one of the following "linkages" does L represent?

Α

5. A polymer formed by addition polymerisation has the following structure:

What is the structure of the monomer?

6. Which one of the following structures correctly represents the linkage between the monomer units in Terylene?

Α

В

D



7. Which one of the following substances is **not** a polymer?

> Α Nylon.

В Poly(ethene).

C Propene. D Starch.

- **8.** Some synthetic products are said to be non-biodegradable. What does this term mean?
  - A Cannot act as catalysts for biological processes.
  - **B** Cannot be obtained from plant or animal matter.
  - **C** Not harmful to living organisms.
  - **D** Not broken down by bacteria.
- **9.** The structure of a monomer is shown below:

$$\begin{array}{ccc}
H & CI \\
C = C \\
F & F
\end{array}$$

Which formula shows two repeating units of the polymer made from this monomer?

- **10.** Which of the following is an example of an addition polymer?
  - A Nylon.

B Poly(ethene).

**C** Terylene.

**D** Starch.

Write your answers to the multiple choice questions in the table below:

1.	2.	3.	4.	5.
6.	7.	8.	9.	10.

### **Short Answer Questions**

#### Question 1.

b)

Terylene is a synthetic polymer which is mainly used for the manufacture of clothing. It is a polyester and hence the clothing should not be washed in a strongly alkaline detergent.

a)	What is meant by the term "polymer"?

Name one other synthetic polymer which is also used for the manufacture of clothing.

**c)** Give the name and the full structural formula of an ester which contains a total of four carbon atoms.

Name: Formula:

d) Name the types of compounds that are formed when an ester is boiled with a strong alkali.

e) Terylene does not have a sharp melting point. What does this suggest about Terylene?

#### Question 2.

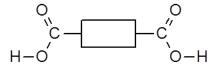
The structure of a polymer is shown below:

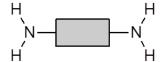
a) i) Name the monomer from which this polymer is made.

	ii) Give the full structu	ral formula of the mo	onomer from which this polymer is made.	
b)	Is this polymer an addition		nsation polymer? Explain your answer.	
c)	Give the empirical formula	of this polymer.		
				•
d)	Calculate the percentage	carbon, by mass, in t	this polymer.	
Qu	estion 3.			
	e following substances have m	nacromolecular struc	tures:	
	Diamond	Poly(propen	ne) Nylon	
	Poly(ethene)	Graphite	Proteins	
	Starch	Terylene	Silica (Sand)	
Ch	oose from the list:			
a)	<b>All</b> of the addition polymers:			
b)	All of the allotropes of carbon:			
c)	All of the synthetic polymers:			
d)	All of the condensation polyr	ners:		
e)	All of the polymers that conta	ain an amide link:		

#### Question 4.

Nylon is a polyamide that can be synthesised from the monomers shown below:





a) Using the monomers shown above, draw the structure of nylon showing three repeating units.

- b) i) Name the structural linkage that joins the monomers in nylon together.
  - ii) Name one group of naturally occurring compounds that have the same structural link as nylon.

.....

c) State the conditions under which nylon can be hydrolysed to reform the original monomers.

.....

d) Sea fishing nets used to be made from natural fibres. Many nets are now made from nylon. Suggest one advantage, other than strength, and one disadvantage of using nylon rather than natural fibres to make fishing nets.

Advantage:

Disadvantage:

.....

• Scan the QR code given below for the answers to this assignment.



http://www.chemist.sg/polymers/polymer\_assignment\_ans.pdf